#### エチレングリコールモノエチルエーテルアセテート のラットを用いた**吸入による** 13 週間毒性試験報告書

試験番号:0743

## **APPENDICES**

#### **APPENDICES**

APPENDIX 1-1	IDENTITY OF ETHYLENE GLYCOL MONOETHYL ETHER ACETATE IN THE 13-WEEK INHALATION STUDY
APPENDIX 1-2	STABILITY OF ETHYLENE GLYCOL MONOETHYL ETHER ACETATE IN THE 13-WEEK INHALATION STUDY
APPENDIX 2	ENVIRONMENTAL CONDITIONS OF INHALATION CHAMBER IN THE 13-WEEK INHALATION STUDY OF ETHYLENE GLYCOL MONOETHYL ETHER ACETATE
APPENDIX 3	METHODS, UNITS AND DECIMAL PLACE FOR HEMATOLOGY AND BIOCHEMISTRY IN THE 13-WEEK INHALATION STUDY OF ETHYLENE GLYCOL MONOETHYL ETHER ACETATE

#### APPENDIX 1 1

# IDENTITY OF ETHYLENE GLYCOL MONOETHYL ETHER ACETATE IN THE 13-WEEK INHALATION STUDY

### IDENTITY OF ETHYLENE GLYCOL MONOETHYL ETHER ACETATE IN THE 13-WEEK INHALATION STUDY

Test Substance

Ethylene glycol monoethyl ether acetate (Wako Pure Chemical

Industries, Ltd.)

Lot No.

: KWJ3548

#### 1. Spectral Data

#### Mass Spectrometry

Instrument

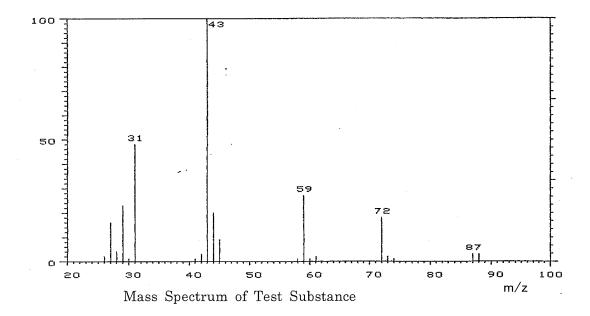
: Hitachi M-80B Mass Spectrometer

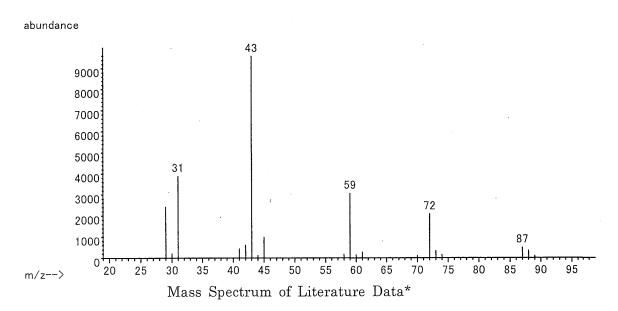
Ionization

: EI (Electron Ionization)

Ionization Voltage

: 70eV





Result: The mass spectrum was consistent with literature spectrum.

(\*McLafferty FW, ed. 1994. Wiley Registry of Mass Spectral Data. 6th ed.

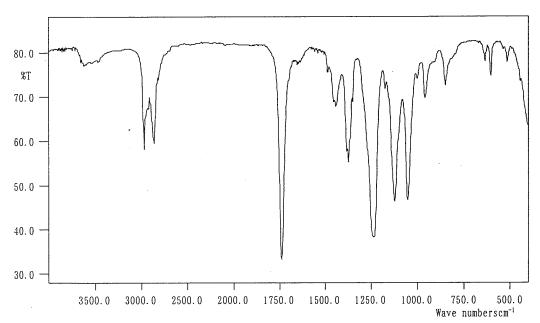
New York, NY John Wiley and Sons.)

#### Infrared Spectrometry

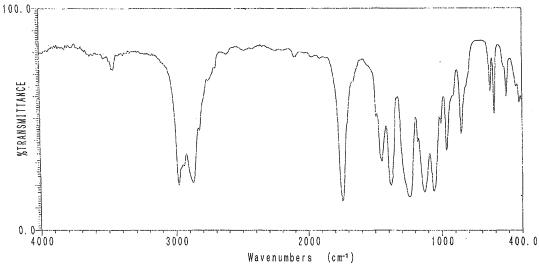
Instrument : Shimadzu FTIR-8200PC Infrared Spectrometer

Cell : KBr Liquid Cell

Resolution : 4 cm<sup>-1</sup>



Infrared Spectrum of Test Substance



Infrared Spectrum of Literature Data\*

Result: The infrared spectrum was consistent with literature spectrum. (\*Performed by Wako Pure Chemical Industries, Ltd.)

2. Conclusion: The test substance was identified as ethylene glycol monoethyl ether acetate by mass spectrum and infrared spectrum.

#### APPENDIX 1 2

# STABILITY OF ETHYLENE GLYCOL MONOETHYL ETHER ACETATE IN THE 13-WEEK INHALATION STUDY

### STABILITY OF ETHYLENE GLYCOL MONOETHYL ETHER ACETATE IN THE 13-WEEK INHALATION STUDY

Test Substance

: Ethylene glycol monoethyl ether acetate (Wako Pure Chemical

Industries, Ltd.)

Lot No.

: KWJ3548

1. Gas Chromatography

Instrument

: Agilent Technologies 5890A Gas Chromatograph

Column

: INNOWAX (  $0.53 \text{ mm} \phi \times 60 \text{ m}$ )

Column Temperature: 140° C

Flow Rate

: 10 mL/min

Detector

: FID (Flame Ionization Detector)

Injection Volume

: 1 µL

Date (date analyzed)	Peak No.	Retention Time (min)	Area (%)
2009.08.25	1	2.880	100
2009.12.08	1	2.875	100

Result: Gas chromatography indicated one major peak (peak No.1) analyzed on 2009.8.25 and one major peak (peak No.1) analyzed on 2009.12.8.

No new trace impurity peak in the test substance analyzed on 2009.12.8 was detected.

2. Conclusion: The test substance was stable for the period that the test substance had been used for the study.

#### APPENDIX 2

## ENVIRONMENTAL CONDITIONS OF INHALATION CHAMBER IN THE 13-WEEK INHALATION STUDY OF ETHYLENE GLYCOL MONOETHYL ETHER ACETATE

## ENVIRONMENTAL CONDITIONS OF INHALATION CHAMBER IN THE 13-WEEK INHALATION STUDY OF ETHYLENE GLYCOL MONOETHYL ETHER ACETATE

Group Name	Temperature $(^{\circ}C)$ Mean $\pm$ S.D.	Humidity $(\%)$ Mean $\pm$ S.D.	Ventilation Rate (L/min) Mean ± S.D.	Air Change (time/h) Mean
Control	$22.4 \pm 0.2$	$55.7 \pm 0.7$	$212.1 \pm 0.9$	12.0
$25~\mathrm{ppm}$	$22.7 \pm 0.3$	$53.4 \pm 1.1$	$212.2 \pm 0.6$	12.0
$50~\mathrm{ppm}$	$22.7 \pm 0.3$	$54.4 \pm 1.2$	$212.4 \pm 0.7$	12.0
100 ppm	$22.5 \pm 0.3$	$53.2 \pm 1.2$	$211.7 \pm 0.7$	12.0
200 ppm	$22.6 \pm 0.3$	$54.3 \pm 1.4$	$212.4 \pm 0.8$	12.0
400 ppm	$22.4 \pm 0.2$	$53.7 \pm 1.4$	$211.9 \pm 0.7$	12.0

#### APPENDIX 3

METHODS, UNITS AND DECIMAL PLACE FOR
HEMATOLOGY AND BIOCHEMISTRY IN THE 13-WEEK
INHALATION STUDY OF
ETHYLENE GLYCOL MONOETHYL ETHER ACETATE

## METHODS, UNITS AND DECIMAL PLACE FOR HEMATOLOGY AND BIOCHEMISTRY IN THE 13-WEEK INHALATION STUDY OF ETHYLENE GLYCOL MONOETHYL ETHER ACETATE

Item	Method	Unit	Decimal place
Hematology			
Red blood cell (RBC)	Light scattering method 1)	$ imes 10^6/\mu\mathrm{L}$	2
Hemoglobin(Hgb)	Cyanmethemoglobin method 1)	g/dL	1
Hematocrit(Hct)	Calculated as RBC×MCV/10 10	%	1
Mean corpuscular volume(MCV)	Light scattering method 1)	fL	1
Mean corpuscular hemoglobin(MCH)	Calculated as Hgb/RBC×10 10	pg	1
Mean corpuscular hemoglobin concentration	Calculated as Hgb/Hct×100 10	g/dL	1
(MCHC)			
Platelet	Light scattering method 1)	$ imes 10^3 / \mu   ext{L}$	0
Reticulocyte	Light scattering method 1)	%	1
Prothrombin time	Quick one stage method 2)	sec	1
Activated partial thromboplastin time (APTT)	Ellagic acid activated method 2)	sec	1
White blood cell(WBC)	Light scattering method 1)	$ imes 10^3 / \mu   m L$	2
Differential WBC	Pattern recognition method 1)	%	0
	(Wright staining)		
Biochemistry			
Total protein(TP)	Biuret method 3)	g/dL	1
Albumin (Alb)	BCG method 3)	g/dL	1
A/G ratio	Calculated as Alb/(TP-Alb) 3)	_	1
T-bilirubin	Azobilirubin method 3)	mg/dL	2
Glucose	GlcK·G-6-PDH method 3)	mg/dL	0
T-cholesterol	CE·COD·POD method 3)	mg/dL	0
Triglyceride	MGLP·GK·GPO·POD method 3)	mg/dL	0
Phospholipid	PLD·ChOD·POD method 3)	mg/dL	0
Aspartate aminotransferase (AST)	JSCC method <sup>3)</sup>	IU/L	0
Alanine aminotransferase (ALT)	JSCC method 3)	IU/L	0
Lactate dehydrogenase (LDH)	JSCC method 3)	IU/L	0
Alkaline phosphatase (ALP)	JSCC method 3)	IU/L	0
$\gamma$ -Glutamyl transpeptidase ( $\gamma$ -GTP)	JSCC method <sup>3)</sup>	IU/L	0
Creatine kinase (CK)	JSCC method 3)	IU/L	0
Urea nitrogen	Urease GLDH method 3)	mg/dL	1
Creatinine	Jaffé method <sup>3)</sup>	mg/dL	1
Sodium	Ion selective electrode method 3)	mEq/L	0
Potassium	Ion selective electrode method 3)	mEq/L	1
Chloride	Ion selective electrode method 3)	mEq/L	0
Calcium	OCPC method 3)	mg/dL	1
Inorganic phosphorus	PNP·XOD·POD method 3)	mg/dL	1

- 1) Automatic blood cell analyzer (ADVIA120 : Siemens Healthcare Diagnostics Inc.)
- 2) Automatic coagulometer (Sysmex CA-5000 : Sysmex Corporation)
- 3) Automatic analyzer (Hitachi 7080: Hitachi, Ltd.)