

アクリル酸のラットを用いた
吸入による 13 週間毒性試験報告書

試験番号 : 0677

APPENDICES

APPENDICES

- APPENDIX 1 1 IDENTITY OF ACRYLIC ACID IN THE 13-WEEK INHALATION STUDY
- APPENDIX 1 2 STABILITY OF ACRYLIC ACID IN THE 13-WEEK INHALATION STUDY
- APPENDIX 2 ENVIRONMENTAL CONDITIONS OF INHALATION CHAMBER IN THE 13-WEEK INHALATION STUDY OF ACRYLIC ACID
- APPENDIX 3 METHODS, UNITS AND DECIMAL PLACE FOR HEMATOLOGY AND BIOCHEMISTRY IN THE 13-WEEK INHALATION STUDY OF ACRYLIC ACID

APPENDIX 1-1

IDENTITY AND IMPURITY OF ACRYLIC ACID
IN THE 13-WEEK INHALATION STUDY

IDENTITY OF ACRYLIC ACID IN THE 13-WEEK INHALATION STUDY

Test Substance : Acrylic acid (Wako Pure Chemical Industries, Ltd.)

Lot No. : DPG1425

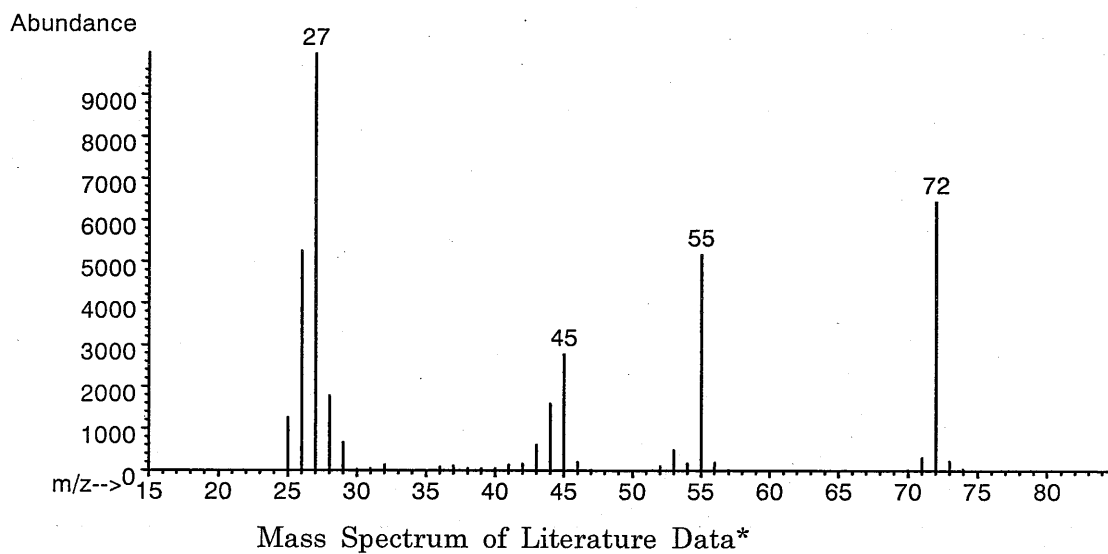
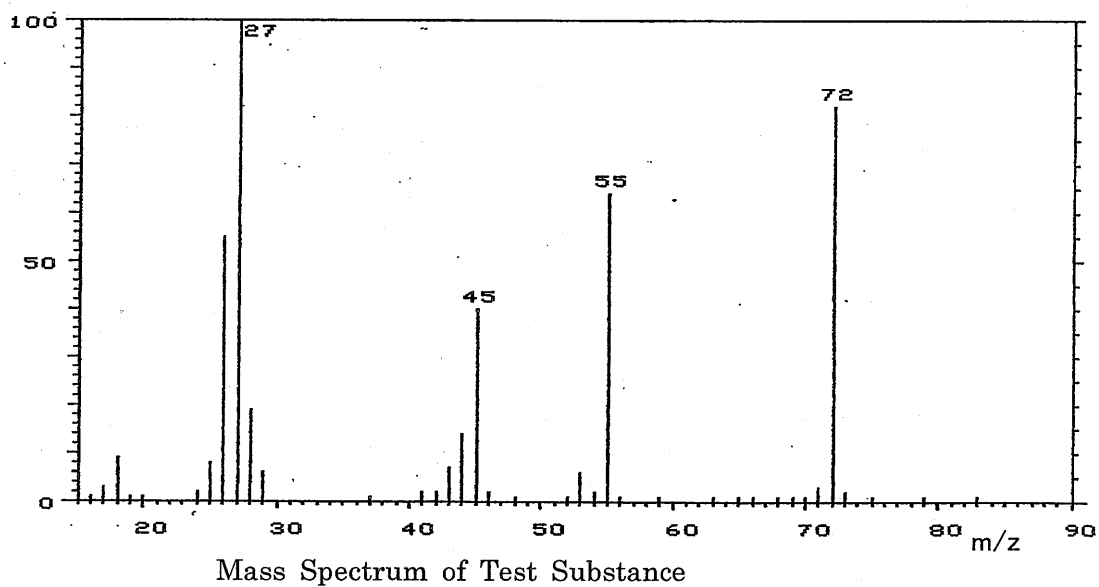
1. Spectral Data

Mass Spectrometry

Instrument : Hitachi M-80B Mass Spectrometer

Ionization : EI (Electron Ionization)

Ionization Voltage : 70eV



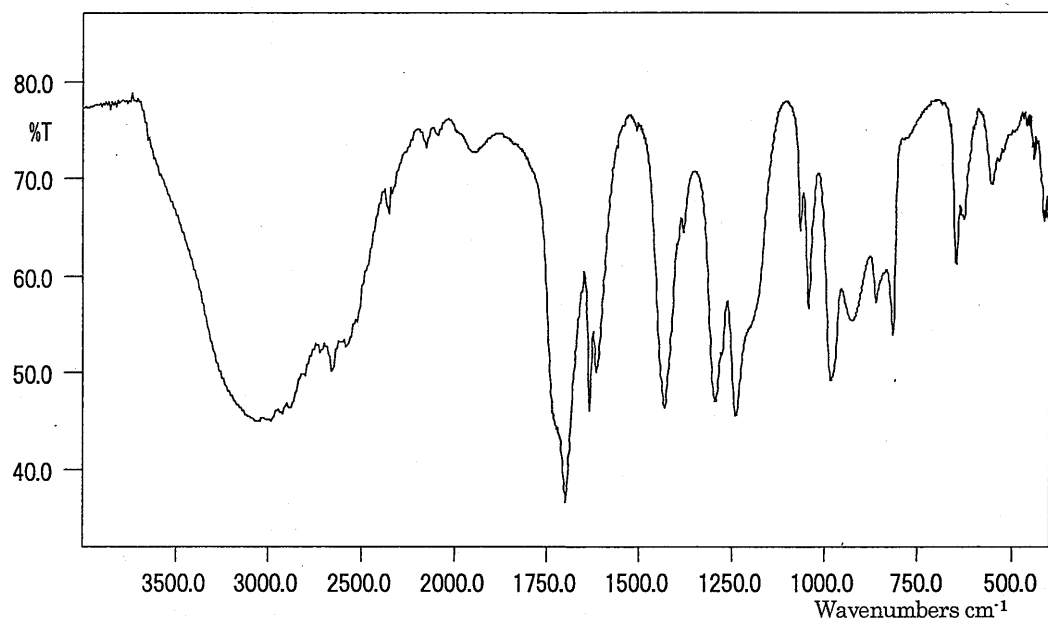
Result: The mass spectrum was consistent with literature spectrum.

(*McLafferty FW, ed. 1994. Wiley Registry of Mass Spectral Data. 6th ed. New York, NY:John Wiley and Sons.)

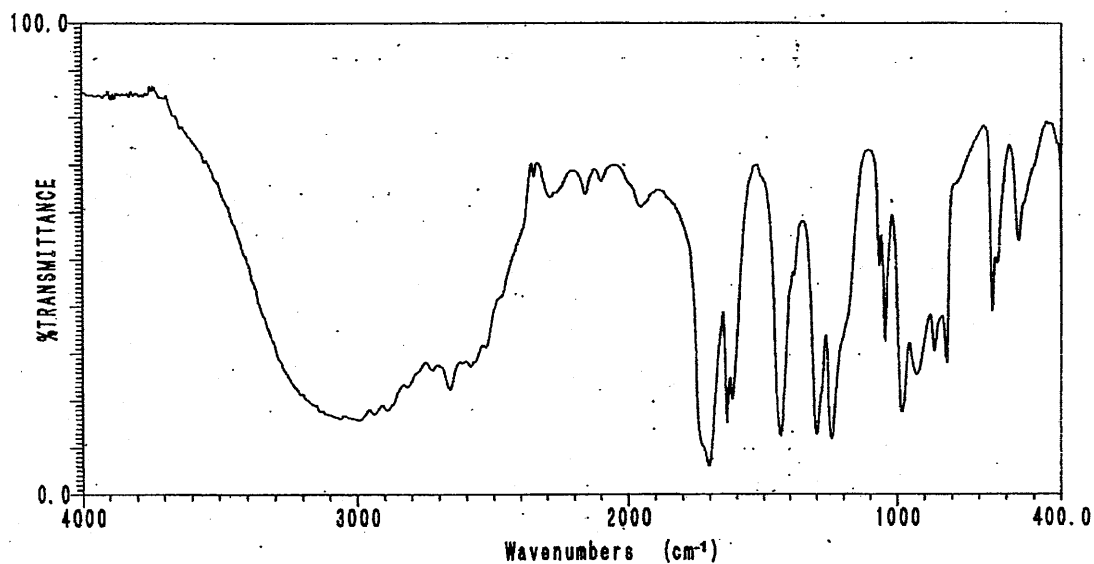
Infrared Spectrometry

Instrument : Shimadzu FTIR-8200PC Infrared Spectrometer

Cell : KBr Liquid Cell

Resolution : 4 cm^{-1} 

Infrared Spectrum of Test Substance



Infrared Spectrum of Literature Data*

Result: The infrared spectrum was consistent with literature spectrum.

(*Performed by Wako Pure Chemical Industries, Ltd.)

2. Conclusion: The test substance was identified as acrylic acid by mass spectrum and infrared spectrum.

APPENDIX 1-2

STABILITY OF ACRYLIC ACID
IN THE 13-WEEK INHALATION STUDY

STABILITY OF ACRYLIC ACID IN THE 13-WEEK INHALATION STUDY

Test Substance : Acrylic acid (Wako Pure Chemical Industries, Ltd.)

Lot No. : DPG1425

1. Gas Chromatography

Instrument : Agilent Technologies 5890A Gas Chromatograph

Column : INNOWAX (0.53 mm ϕ \times 60 m)

Column Temperature: 150° C

Flow Rate : 20 mL/min

Detector : FID (Flame Ionization Detector)

Injection Volume : 1 μ L

Date (date analyzed)	Peak No.	Retention Time (min)	Area (%)
2007.01.22	1	2.882	100
2007.05.07	1	2.892	100

Result: Gas chromatography indicated one major peak (peak No.1) analyzed on 2007.1.22 and one major peak (peak No.1) analyzed on 2007.5.7.
No new trace impurity peak in the test substance analyzed on 2007.5.7 was detected.

2. Conclusion: The test substance was stable for the period that the test substance had been used for the study.

APPENDIX 2

ENVIRONMENTAL CONDITIONS OF INHALATION CHAMBER
IN THE 13-WEEK INHALATION STUDY OF
ACRYLIC ACID

ENVIRONMENTAL CONDITIONS OF INHALATION CHAMBER IN THE 13-WEEK
INHALATION STUDY OF ACRYLIC ACID

Group Name	Temperature (°C) Mean ± S.D.	Humidity (%) Mean ± S.D.	Ventilation Rate (L/min) Mean ± S.D.	Air Change (time/h) Mean
Control	22.9 ± 0.2	56.0 ± 0.3	212.3 ± 0.2	12.0
10 ppm	22.8 ± 0.1	51.3 ± 1.5	212.4 ± 0.3	12.0
24 ppm	22.9 ± 0.1	51.0 ± 1.9	212.4 ± 0.3	12.0
60 ppm	22.7 ± 0.1	50.0 ± 2.3	212.4 ± 0.3	12.0
150 ppm	22.8 ± 0.1	46.5 ± 2.8	212.3 ± 0.3	12.0
200 ppm	22.6 ± 0.1	46.0 ± 3.1	212.2 ± 0.2	12.0

APPENDIX 3

METHODS, UNITS AND DECIMAL PLACE FOR
HEMATOLOGY AND BIOCHEMISTRY IN THE 13-WEEK
INHALATION STUDY OF ACRYLIC ACID

METHODS, UNITS AND DECIMAL PLACE FOR HEMATOLOGY AND BIOCHEMISTRY
IN THE 13-WEEK INHALATION STUDY OF ACRYLIC ACID

Item	Method	Unit	Decimal place
Hematology			
Red blood cell (RBC)	Light scattering method ¹⁾	$\times 10^6/\mu\text{L}$	2
Hemoglobin(Hgb)	Cyanmethemoglobin method ¹⁾	g/dL	1
Hematocrit(Hct)	Calculated as $\text{RBC} \times \text{MCV} / 10$ ¹⁾	%	1
Mean corpuscular volume(MCV)	Light scattering method ¹⁾	fL	1
Mean corpuscular hemoglobin(MCH)	Calculated as $\text{Hgb} / \text{RBC} \times 10$ ¹⁾	pg	1
Mean corpuscular hemoglobin concentration (MCHC)	Calculated as $\text{Hgb} / \text{Hct} \times 100$ ¹⁾	g/dL	1
Platelet	Light scattering method ¹⁾	$\times 10^3/\mu\text{L}$	0
Reticulocyte	Light scattering method ¹⁾	%	1
Prothrombin time	Quick one stage method ²⁾	sec	1
Activated partial thromboplastin time (APTT)	Ellagic acid activated method ²⁾	sec	1
White blood cell(WBC)	Light scattering method ¹⁾	$\times 10^3/\mu\text{L}$	2
Differential WBC	Pattern recognition method ³⁾ (Wright staining)	%	0
Biochemistry			
Total protein(TP)	Biuret method ⁴⁾	g/dL	1
Albumin (Alb)	BCG method ⁴⁾	g/dL	1
A/G ratio	Calculated as $\text{Alb} / (\text{TP} - \text{Alb})$ ⁴⁾	-	1
T-bilirubin	Azobilirubin method ⁴⁾	mg/dL	2
Glucose	GlcK·G-6-PDH method ⁴⁾	mg/dL	0
T-cholesterol	CE·COD·POD method ⁴⁾	mg/dL	0
Triglyceride	MGLP·GK·GPO·POD method ⁴⁾	mg/dL	0
Phospholipid	PLD·ChOD·POD method ⁴⁾	mg/dL	0
Aspartate aminotransferase (AST)	JSCC method ⁴⁾	IU/L	0
Alanine aminotransferase (ALT)	JSCC method ⁴⁾	IU/L	0
Lactate dehydrogenase (LDH)	SFBC method ⁴⁾	IU/L	0
Alkaline phosphatase (ALP)	GSCC method ⁴⁾	IU/L	0
γ -Glutamyl transpeptidase (γ -GTP)	JSCC method ⁴⁾	IU/L	0
Creatine kinase (CK)	JSCC method ⁴⁾	IU/L	0
Urea nitrogen	Urease·GLDH method ⁴⁾	mg/dL	1
Creatinine	Jaffé method ⁴⁾	mg/dL	1
Sodium	Ion selective electrode method ⁴⁾	mEq/L	0
Potassium	Ion selective electrode method ⁴⁾	mEq/L	1
Chloride	Ion selective electrode method ⁴⁾	mEq/L	0
Calcium	OCPC method ⁴⁾	mg/dL	1
Inorganic phosphorus	PNP·XOD·POD method ⁴⁾	mg/dL	1

1) Automatic blood cell analyzer (ADVIA120 : Bayer Corporation)

2) Automatic coagulometer (Sysmex CA-5000 : Sysmex Corporation)

3) Automatic blood cell differential analyzer (MICROX HEG-120NA : OMRON Corporation)

4) Automatic analyzer (Hitachi 7080 : Hitachi, Ltd.)